

**PCT**WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification <sup>6</sup> :</b> <b>C07C 323/59 // A61K 31/195, 31/22, 31/225</b>	<b>A1</b>	<b>(11) International Publication Number:</b> <b>WO 97/48679</b> <b>(43) International Publication Date:</b> 24 December 1997 (24.12.97)
<b>(21) International Application Number:</b> PCT/SE97/01069 <b>(22) International Filing Date:</b> 17 June 1997 (17.06.97) <b>(30) Priority Data:</b> 9602416-1 18 June 1996 (18.06.96) SE <b>(71) Applicant (for all designated States except US):</b> ASTRA AKTIEBOLAG (publ) [SE/SE]; S-151 85 Södertälje (SE). <b>(72) Inventors; and</b> <b>(75) Inventors/Applicants (for US only):</b> JAKUPOVIC, Edib [SE/SE]; Smultronvägen 7, S-155 00 Nykvarn (SE). TENEBERG, Eric [SE/SE]; Längsjöhöjden 58, S-125 51 Älvsjö (SE). <b>(74) Agent:</b> ASTRA AKTIEBOLAG; Patent Dept., S-151 85 Södertälje (SE).		<b>(81) Designated States:</b> AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).  <b>Published</b> <i>With international search report.</i>
<b>(54) Title:</b> NEW FORMS OF ORGANIC SALTS OF N'-DIACETYLCYSTINE  <b>(57) Abstract</b>  New hydrated salts of N,N'-diacetylcystine are described, as well as processes for the preparation thereof.		

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

## NEW FORMS OF ORGANIC SALTS OF N,N-DIACETYLCYSTINE

The present invention relates to new hydrates of salts of N,N-diacetylcystine and to processes for the preparation thereof.

Background of the Invention

N-acetyl-L-cysteine is a well-known compound which is used as a therapeutic agent against chronic obstructive pulmonary diseases and chronic bronchitis. Although the first patent was filed in 1964 (GB 954268), the mechanism of action of the compound has not been established. It is also known that the corresponding disulphide of N-acetyl-L-cysteine, i.e., N,N-diacetyl-L-cystine, L-DiNAC, acts as a potent immunostimulator (SE patent application No. 9002067-8), showing an activity comparable to contemporary immunostimulants such as sodium diethyl dithiocarbamate or 2,2'-dithiobisethanol.

It has recently been found that certain salts of DiNAC with organic bases exhibit a favourable combination of non-hygroscopicity and crystallinity which permits the isolation and formulation of these salts in solid form. They have the advantages of ease of crystallisation, non-hygroscopicity and chemical stability, as well as the immunomodulating activity of DiNAC, and are thus medically useful. These salts are described in WO 93/11104.

Salts composed of an organic base and N,N-diacetylcystine (DiNAC) are generally prepared by mixing DiNAC and the organic base, as defined above, each dissolved or dispersed in a solvent or solvent mixture. Solvents, such as water, alcohols, glycols, ketones, amides, sulphoxides or other polar solvents may be used; solvent mixtures may also be used. The salt either precipitates directly from the reaction mixture, or is obtained by the addition of a less polar solvent, by evaporation, or by lyophilisation. The reaction is performed at elevated temperature or room temperature, depending on the solubility in the medium. Alternatively, the salt can be prepared by oxidation of the appropriate N-acetyl

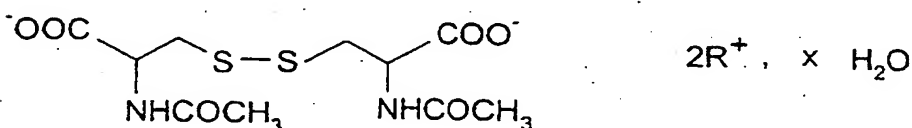
cysteine salt in an aqueous or alcoholic solution, followed by precipitation as above. The oxidation may be effected either chemically, using, e.g., hydrogen peroxide or a halogen, or electrochemically.

5 The above methods provide the organic salts of DiNAC in anhydrous form.

#### Disclosure of the invention

We have now prepared novel hydrates of certain salts of N,N'-diacetylcystine.

10 The invention, in one aspect, provides hydrated salts of N,N'-diacetylcystine, having the formula



in which

the N,N'-diacetylcystine is the D-, L-, or meso form, or any mixture thereof,

15  $\text{R}^+$  is the protonated form of lysine, ammonia or N-benzyl-2-phenylethylamine, and  $x$  is 1, 1.5, or 2, designating the monohydrate, sesquihydrate, and dihydrate, respectively, provided that when, and only when,  $\text{R}^+$  is the protonated form of N-benzyl-2-phenylethylamine,  $x$  is 1.5.

20 The N,N'-diacetylcystine is preferably the L-form. The lysine is preferably L-lysine.

In another aspect, the invention provides a process for obtaining the hydrated salts. The process includes oxidising N-acetyl cysteine in the presence of lysine, ammonia or N-benzyl-2-phenylethylamine, or the protonated forms thereof, in an aqueous or alcoholic  
25 solution, preferably in the presence of a catalytic amount of an alkali metal hydroxide, and

crystallising the hydrate using a solvent comprising an alcohol or a ketone, preferably ethanol or acetone.

When the hydrated salt is a dihydrate, or the sesquihydrate, the crystallisation is carried out using a solvent comprising an alcohol, preferably ethanol. When the hydrated salt is the lysinium monohydrate, the crystallisation is carried out using a solvent comprising an alcohol, preferably ethanol, and when the hydrated salt is the ammonium monohydrate, the crystallisation is carried out using a solvent comprising a ketone, preferably acetone.

When the hydrated salt is a dihydrate, or the sesquihydrate, the crystallisation is carried out using a solvent comprising an alcohol, preferably ethanol. When the hydrated salt is the lysinium monohydrate, the crystallisation is carried out using a solvent comprising an alcohol, preferably ethanol, and when the hydrated salt is the ammonium monohydrate, the crystallisation is carried out using a solvent comprising a ketone, preferably acetone.

Specifically, the hydrated forms of di-lysinium- N,N'-diacetylcystinate may be prepared for example by oxidising N-acetyl cysteine in an aqueous or alcoholic solution in the presence of lysine or a salt containing the protonated form of lysine, and a catalytic amount of an alkali metal hydroxide, and then

- a) to obtain the monohydrated lysine salt, crystallising the product by adding an aqueous solution thereof to ethanol, preferably at a temperature of about 45°C;
- b) to obtain the dihydrated salt, crystallising the product by adding ethanol to an aqueous solution thereof at a temperature of about 60°C or below, or by adding an aqueous solution of the product to a mixture of ethanol and water at a temperature of below about 60°C, for example at about 20°C or below.

The monohydrate form of the lysine salt of N,N'-diacetylcystine is metastable, while the dihydrate form is stable and especially suitable for use in inhalation devices.

The hydrated forms of di-ammonium- N,N'-diacetylcystinate may be prepared for example by oxidising N-acetyl cysteine in an aqueous solution in the presence of ammonia or ammonium hydroxide, or in the presence of another salt containing the protonated form of ammonia and a catalytic amount of an alkali metal hydroxide, and

a) to obtain the monohydrated salt, crystallising the product by adding acetone  
b) to obtain the dihydrated salt, crystallising the product by adding ethanol to an aqueous solution thereof.

Crystallisation of the dihydrated ammonium salt as above is preferably carried out at a temperature of for example between about 40°C and about 80°C, preferably between about 40°C and about 78°C, for example at about 78°C.

An alternative method of crystallising the dihydrated ammonium salt involves adding the oxidised product to a mixture of ethanol and water.

Di- N-benzyl-2-phenylethylamine sesquihydrate may be prepared for example by oxidising N-acetyl cysteine in an aqueous or alcoholic solution in the presence of N-benzyl-2-phenylethylamine or a salt containing the protonated form of N-benzyl-2-phenylethylamine, and a catalytic amount of an alkali metal hydroxide, and then adding ethanol and standing at 0-5°C to crystallise.

By "catalytic amount" is meant a trace amount of alkali metal hydroxide; preferably at least 0.001 mole equivalent compared to the N-acetyl-L-cysteine is used, or up to 0.01 or 0.1 mole equivalent. Preferably no more than 0.5 mole equivalent is used. In one embodiment of the invention, about 0.1 mole equivalent of the alkali metal hydroxide is used.

Suitable alkali metal hydroxides are, e.g., sodium, potassium and lithium hydroxide.

The oxidation may be effected either chemically, using, e.g., hydrogen peroxide or halogen as oxidising agent, or electrochemically.

The hydrated salts of the present invention have immunomodulating activity and may be used for example in the treatment of diseases where an anergy of the immune response or an aberrant immune response or an ineffective host response can be suspected. Among such diseases are included the diseases listed in WO 93/11104, incorporated herein by reference. For example, diseases which may benefit from treatment with the present hydrated salts include chronic bronchitis, malignant diseases and chronic infections. The ability of the hydrated salts to modulate immune responses may be illustrated in the animal delayed type hypersensitivity (DTH) test in the mouse, as described in WO 93/11104. The present hydrated salts will also be useful in the treatment of chronic hepatitis B and/or C infections, as may be illustrated by a strengthening of the TH1-type response in the Leishmania model, as described for example by Connell et al ((N. Connell, E. Medina-Acosta, W. McMaster, B. Bloom and D. Russell, 1993, Effective immunisation against cutaneous leishmaniasis with recombinant bacilli Calmette-Guerin expressing the Leishmania surface proteinase p63, Proc. Natl. Acad. Sci. USA, 90:11473).

The new hydrated salts can be formulated for administration by inhalation, for example from a dry powder inhaler or from a pressurised metered dose inhaler (pMDI); alternatively, they can be formulated for oral, topical, or parenteral use. The formulations may include a pharmaceutically acceptable carrier.

The hydrated salts of the present invention can be included in different dosage forms, e.g., dry powders, aerosols, tablets, coated tablets, gelatine capsules and solutions.

For the preparation of a formulation for inhalation from a dry powder inhaler, the hydrated salts of the present invention may be combined with for example a pharmaceutically acceptable diluent or carrier and provided in the form of inhalable particles.

For the preparation of a formulation for inhalation from a pMDI the hydrated salts of the present invention may be dissolved or suspended in a suitable propellant optionally together with a co-solvent and/or one or more pharmaceutically acceptable surfactants or other excipients.

For the preparation of tablets, coated tablets and gelatine capsules the hydrated salts of the present invention can be combined with pharmaceutically acceptable materials, e.g., lactose, starch, dicalcium phosphate, microcrystalline cellulose, polyvinylpyrrolidone, gelatine, cellulose derivatives, colloidal silicone dioxide, talc and stearic acid or its salts.

For the preparation of oral solutions suitable excipients are water, saccharose, glucose, sorbitol, fructose and xylitol.

The dosage forms can besides mentioned excipients contain preservatives, stabilisers, viscosity regulating agents, emulsifiers, sweetening agents, colouring agents, flavouring agents, tonicity regulating agents, buffers or antioxidants. They can also contain other therapeutically valuable substances.

The invention is illustrated by the following non-limiting examples.

#### Example 1

##### *Di-L-lysinium-N,N'-diacetyl-L-cystinate monohydrate*

N-acetyl-L-cysteine (50.0 g, 1.0 eq), L-lysine monohydrate (50.5 g, 1.0 eq), potassium hydroxide (1.6 g, 0.08 eq) and purified water (75 mL) were mixed and stirred until a clear solution was achieved. Hydrogen peroxide, (35% solution, 0.5 eq) was then added dropwise, with the temperature kept between about 20°C and about 40°C during the addition. The solution was heated to 45°C, and ethanol (1500 mL) was added. After



stirring at 45°C for one hour, the crystals were filtered off, giving 71 g (70%) of the title substance.

The  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR spectra showed the following peaks:

$^1\text{H}$ -NMR ( $\text{D}_2\text{O}$ ):  $\delta$ 4.50 (dd, 2H), 3.75 (t, 2H), 3.26 (dd, 2H), 3.03 (t, 4H), 2.96 (dd, 2H), 2.06 (s, 6H), 1.84-1.96 (m, 4H), 1.73 (quintet, 4H), 1.35-1.60 (m, 4H).

$^{13}\text{C}$ -NMR ( $\text{D}_2\text{O}$ ):  $\delta$ 177.6, 175.5, 174.4, 67.3, 55.2, 54.9, 40.7, 39.8, 30.7, 27.2, 22.7, 22.2.

Powder X-ray diffractograms of the product, measured from 1 to 40° in  $2\theta$  showed the following peaks, which are characteristic of the monohydrate:

	Angle [°2θ]	d-value CuKα <sup>1</sup> [Å]	Intensity
15	4.68	18.87	strong
	5.30	16.66	strong
	6.54	13.50	very strong
	7.50	11.79	very strong
20	10.56	8.37	strong
	13.97	6.33	strong
	14.24	6.21	strong
	18.66	4.75	strong
	19.46	4.56	strong
25	20.15	4.40	very strong
	21.17	4.19	strong
	22.33	3.98	strong
	22.59	3.93	strong
	22.81	3.90	strong
30	23.05	3.86	strong

23.38 3.80 strong

23.80 3.74 strong

Karl-Fischer-titration showed 2.8% (w/w) of water which is equivalent to the theoretical value.

### Example 2

#### *Di-L-lysinium-N,N'-diacetyl-L-cystinate dihydrate*

N-acetyl-L-cysteine (50.0 g, 1.0 eq), L-lysine monohydrate (50.5 g, 1.0 eq), potassium hydroxide (1.6 g, 0.08 eq) and purified water (75 mL) were mixed and stirred until a clear solution was achieved. Hydrogen peroxide, (35% solution, 0.5 eq) was then added dropwise, with the temperature kept between about 20°C and about 40°C during the addition.

The dihydrate was then crystallised from the above solution, using each of the following methods:

1). The solution was added to 800 ml of ethanol at 60°C. After 3 h of stirring, the crystals were filtered off, giving 74 g (70%) of the title substance.

2). The solution was added to 800 ml of ethanol, containing 10% of water, at 20°C. After 10 h of stirring, the crystals were filtered off, giving 66.5 g (63%) of the title substance.

The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR were identical to those of the monohydrate.

Powder X-ray diffractograms of the product measured from 1 to 40° in 2θ show the following peaks, characteristic of the dihydrate:

Angle [°2θ]	d-value CuKα [Å]	Intensity
----------------	------------------------	-----------

9.44	9.36	very strong
------	------	-------------

	10.93	8.09	strong
	16.70	5.31	strong
	17.38	5.10	strong
	17.62	5.03	strong
5	17.99	4.93	very strong
	20.01	4.43	strong
	21.95	4.05	strong
	22.66	3.92	strong
	24.11	3.69	strong
10	24.34	3.65	strong
	24.64	3.61	strong
	25.27	3.52	very strong
	25.93	3.43	strong

15 Karl-Fischer-titration showed 5.5% (w/w) of water which is equivalent to the theoretical value.

### Example 3

#### *Di-ammonium-N,N'-diacetyl-L-cystinate, monohydrate*

20 N-acetyl-L-cysteine (50g, 1.0 eq.) and purified water (42mL) were stirred for 15 minutes. Ammonium hydroxide (25% solution, 24 mL, 1.15 eq.) was added dropwise over 20 minutes. To the resulting clear solution was added hydrogen peroxide (14.6 mL, 1.0 eq) over a period of 30 minutes, maintaining a temperature not over 50°C. Acetone (190 mL) was added slowly (over 30 minutes), resulting in crystallisation. The slurry was allowed to  
25 stand with stirring at 0°C for 16h and the crystals were then filtered off, washed with acetone (10 mL) and dried, giving 28.8 g (51%) of the title substance.

The XRD values of the monohydrate were as follows:

Angle [°2 $\theta$ ]	d-value CuK $\alpha$ <sup>1</sup> [Å]	Intensity
6.00	14.73	weak
9.31	9.49	very weak
11.48	7.71	medium
11.99	7.37	very strong
13.97	6.34	very weak
18.01	4.92	medium
18.51	4.79	very weak
18.77	4.73	very weak
19.39	4.58	weak
20.57	4.31	weak
22.21	4.00	weak
22.78	3.90	weak
23.04	3.86	medium
23.86	3.73	weak
25.15	3.54	medium
25.91	3.44	weak
27.71	3.22	very weak
28.14	3.17	very weak
28.84	3.09	very weak
30.24	2.95	weak
33.57	2.67	very weak
34.16	2.62	very weak
34.85	2.57	very weak
36.45	2.46	very weak

Example 4*Di-ammonium-N,N'-diacetyl-L-cystinate dihydrate*

N-acetyl-L-cysteine (50g, 1.0 eq.) and purified water (42mL) were stirred for 15 minutes. Ammonium hydroxide (25% solution, 24 mL, 1.15 eq.) was added dropwise over 20 minutes. To the resulting clear solution was added hydrogen peroxide (14.6 mL, 1.0 eq) over a period of 30 minutes, maintaining a temperature not over 50°C. The solution was then slowly added to ethanol (170 mL) at 40-78°C, over 45 minutes. After the addition the temperature was decreased to 0-5°C, resulting in precipitation of crystals. Filtration, washing with ethanol (20 mL) and drying gave 12.9 g (22%) of the title substance.

The XRD values of the monohydrate were as follows:

Angle [°2θ]	d-value CuKα <sup>1</sup> [Å]	Intensity
6.28	14.06	weak
7.40	11.94	weak
7.67	11.52	medium
11.63	7.61	medium
13.61	6.50	weak
13.88	6.38	very weak
14.61	6.06	very weak
14.86	5.96	very weak
15.02	5.90	very weak
15.60	5.68	very weak
15.87	5.58	weak
17.15	5.17	weak
18.69	4.75	medium

19.13	4.64	weak
22.20	4.00	medium
23.96	3.71	very weak
24.35	3.65	very weak

### Example 5

#### 5 *Di-N-benzyl-2-phenylethylammonium-N,N'-diacetyl-L-cystinate, sesquihydrate*

N-acetyl-L-cysteine (50 g, 1.0 eq), purified water (75 mL), potassium hydroxide (1.8 g; 0.1 eq) and N-benzyl-2-phenylethylamine (64.6 g, 2.0 eq) were mixed at approx. 25°C. To the mixture was added hydrogen peroxide with cooling to maintain the reaction temperature at  
10 25-30°C. Ethanol (177 mL) was then added and the resulting mixture was allowed to stand overnight at 0-5°C to crystallise. The obtained crystals were filtered off, washed with ethanol (50 mL) and dried to give 17.2 g (10%) of the title compound.

The XRD values for the sesquihydrate were as follows.

15

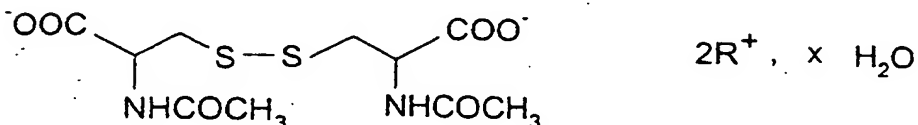
Angle [°2θ]	d-value CuKα <sup>1</sup> [Å]	Intensity
6.20	14.26	strong
9.63	9.18	strong
10.01	8.83	medium
12.42	7.12	very weak
12.78	6.92	weak
13.61	6.50	medium
14.92	5.93	weak

15.22	5.82	weak
15.83	5.59	very weak
16.54	5.36	medium
17.35	5.11	weak
18.67	4.75	strong
19.26	4.60	strong
20.49	4.33	strong
20.83	4.26	medium
21.47	4.14	strong
21.82	4.07	medium
22.05	4.03	medium
23.18	3.83	medium
23.85	3.73	very weak
24.23	3.67	medium
25.68	3.47	very weak
26.01	3.42	very weak
26.11	3.41	very weak
27.89	3.20	medium
29.16	3.06	weak
31.08	2.88	very weak
33.12	2.70	very weak
33.49	2.67	very weak
35.19	2.55	very weak
36.20	2.48	very weak
37.44	2.40	very weak

The X-ray diffractograms also contained several medium, weak and very weak peaks which have been omitted for clarity.

## CLAIMS

1. A hydrated salt of N,N'-diacetylcystine, wherein the hydrated salt has the formula



in which

the N,N'-diacetylcystine is the D-, L-, or meso form, or any mixture thereof,

R<sup>+</sup> is the protonated form of lysine, ammonia or N-benzyl-2-phenylethylamine, and

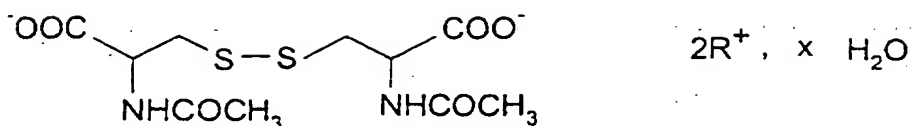
x is 1, 1.5, or 2, designating the monohydrate, sesquihydrate, and dihydrate, respectively, provided that when, and only when, R<sup>+</sup> is the protonated form of N-benzyl-2-phenylethylamine, x is 1.5.

2. A hydrated salt as claimed in claim 1, in which the N,N'-diacetylcystine is the L-form.

3. A hydrated salt as claimed in claim 1, wherein said salt is selected from the group consisting of di-L-lysinium-N,N'-diacetyl-L-cystinate monohydrate and di-L-lysinium-N,N'-diacetyl-L-cystinate dihydrate.

4. A process for the preparation of a hydrated salt of N,N'-diacetylcystine, wherein the hydrated salt has the formula





in which

the N,N'-diacetylcystine is the D-, L-, or meso form, or any mixture thereof,

$\text{R}^+$  is the protonated form of lysine, ammonia or N-benzyl-2-phenylethylamine, and

x is 1, 1.5, or 2, designating the monohydrate, sesquihydrate, and dihydrate, respectively, provided that when, and only when,  $\text{R}^+$  is the protonated form of N-benzyl-2-phenylethylamine, x is 1.5,

comprising

(a) providing a first solution comprising i) N-acetyl cysteine, ii) lysine, ammonia or N-benzyl-2-phenylethylamine or the protonated forms thereof, iii) a solvent, wherein the solvent is water or alcohol, and iv) optionally a catalytic amount of an alkali metal hydroxide;

(b) applying an oxidant to the solution; and

(c) (i) adding the first solution to a second solution, wherein the second solution comprises ethanol,  $\text{R}^+$  is the protonated form of lysine, and x is 1; or

(ii) adding to the first solution a second solution, wherein the second solution comprises acetone,  $\text{R}^+$  is the protonated form of ammonia, and x is 1; or

(iii) adding to the first solution a second solution comprising ethanol, or adding the first solution to a second solution comprising a mixture of ethanol and water, wherein  $\text{R}^+$  is the protonated form of lysine or ammonia and x is 2; or

(iv) adding to the first solution a second solution, wherein the second solution comprises ethanol,  $\text{R}^+$  is the protonated form of N-benzyl-2-phenylethylamine, and x is 1.5.

5. The process of claim 4, wherein step (c)(i) is followed.
6. The process of claim 4, wherein step (c)(ii) is followed.
7. The process of claim 4, wherein step (c)(iii) is followed.
8. The process of claim 4, wherein step (c)(iv) is followed.
9. A therapeutic composition comprising as active ingredient a hydrated salt as claimed in any one of claims 1-3.
10. A therapeutic composition as claimed in claim 9, formulated for administration by inhalation from a dry powder inhaler.
11. Use of a hydrated salt as claimed in any one of claims 1-3, in the preparation of a medicament with immunomodulating activity.
12. Use of a hydrated salt as claimed in any one of claims 1-3, in the preparation of a medicament for the treatment of chronic bronchitis.
13. Use of a hydrated salt as claimed in any one of claims 1-3, in the preparation of a medicament for the treatment of malignant diseases.
14. Use of a hydrated salt as claimed in any one of claims 1-3, in the preparation of a medicament for the treatment of chronic infections.
15. Use of a hydrated salt as claimed in any one of claims 1-3, in the preparation of a medicament for the treatment of chronic hepatitis B and/or C infections.

16. Di-L-lysinium-N,N'-diacetyl-L-cystinate monohydrate.

17. Di-L-lysinium-N,N'-diacetyl-L-cystinate dihydrate.

5 18. Di-ammonium-N,N'-diacetyl-L-cystinate monohydrate.

19. Di-ammonium-N,N'-diacetyl-L-cystinate dihydrate.

20. Di-N-benzyl-2-phenylethylaminium-N,N'-diacetyl-L-cystinate sesquihydrate.

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/SE 97/01069

## A. CLASSIFICATION OF SUBJECT MATTER

IPC6: C07C 323/59 // A61K 31/195, A61K 31/22, A61K 31/225  
According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC6: C07C, A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

SE,DK,FI,NO classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

CA, WPI

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 9311104 A1 (AB ASTRA), 10 June 1993 (10.06.93), page 3, line 1 - line 9; page 6, line 25 - line 29; page 16; page 17, line 26; page 18, line 9 - line 19; page 19; the claims	1-20

☐ Further documents are listed in the continuation of Box C.

☒ See patent family annex.

\* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

18 Sept 1997

Date of mailing of the international search report

24 -09- 1997

Name and mailing address of the ISA/  
Swedish Patent Office  
Box 5055, S-102 42 STOCKHOLM  
Facsimile No. +46 8 666 02 86

Authorized officer

Gerd Strandell

Telephone No. +46 8 782 25 00

# INTERNATIONAL SEARCH REPORT

Information on patent family members

01/09/97

International application No.

PCT/SE 97/01069

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9311104 A1	10/06/93	AP 350 A	09/08/94
		AP 9200448 D	00/00/00
		AT 141913 T	15/09/96
		AU 657827 B	23/03/95
		AU 3099692 A	28/06/93
		CA 2123516 A	10/06/93
		CN 1073683 A	30/06/93
		CZ 9401297 A	18/01/95
		DE 69213256 D,T	23/01/97
		EP 0621862 A,B	02/11/94
		SE 0621862 T3	
		ES 2090713 T	16/10/96
		FI 942504 A	27/05/94
		HR 921322 A	31/10/96
		HU 70851 A	28/11/95
		HU 9401473 D	00/00/00
		JP 7501539 T	16/02/95
		NO 941968 A	26/05/94
		NZ 245077 A	27/04/95
		PL 170787 B	31/01/97
		PL 171336 B	30/04/97
		SE 9103572 D	00/00/00
		SK 63294 A	08/02/95
		US 5385904 A	31/01/95
		US 5650538 A	22/07/97
		ZA 9208739 A	09/06/93

